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PHD (Chemical) ADVANCED ORGANIC CHEMISTRY

Time: 3 Hrs. Max. Marks: 100

INSTRUCTIONS TO CANDIDATES:

- 1. Attempt any FIVE questions out of EIGHT questions.
- 2. Each question carries TWENTY marks.
- Q1 i) What are electrocyclic reactions? Drawing correlation diagrams, discuss disrotatory and conrotatory interconversion of cyclobutene and butadiene.
 - ii) Discuss the Claisen and Cope rearrangements with mechanism.
- Q2 i) What are suprafacial and antarafacial processes in sigmatropic rearrangements? Illustrate these processes by examining a suprafacial 1, 5-sigmatropic shift of hydrogen in which hemolytic cleavage results in the production of a hydrogen atom and pentadienyl radicals. Explain with energy level diagram.
 - ii) What are the products of concerted cycloaddition of ethylene, addition being suprafacial on both components? Is it symmetry allowed under thermal or photochemical condition? Explain.
- Q3 i) Discuss with mechanism for hydrosilylation and hydroboration reactions.
 - ii) How Suzuki reaction is different from Sonogashira reaction. Explain with mechanism.
- Q4 i) Explain how quaternary salts and crown ethers are helpful for organic reactions in phase transfer catalysts.
 - ii) How transition metals are used for the cleavage of C-H bonds? Explain with suitable examples. Draw the mechanism wherever applicable.
- Q5 i) Discuss the regiochemistry and stereochemistry in Diels Alder reaction.
 - ii) What do you understand by 1,3-dipolar cycloaddition reactions? Explain dipolar reaction with the help of molecular orbital diagrams.
- O6 i) Show the mechanism of (a) povarov and (b) aza-Diels-Alder reaction.
 - ii) Discuss hetero Diels-Alder reaction of 4 + 2 cycloaddition reaction.
- O7 i) Discuss the multicomponent cycloaddition type of reaction of 5 + 2 + 1.
 - ii) Explain with suitable example for 5 + 2 type of cycloaddition multicomponent reaction.
- Q8 i) How transition metal salts are used for the multicomponent cycloaddition reaction? Support your answer with examples.
 - ii) Discuss with suitable example for 3 + 2 + 2 multicomponent cycloaddition reaction.

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