Roll No.

Total No. of Pages: 02

Total No. of Questions: 08

M.Sc. (Ph. Chem) (2018 Batch) (Sem.-1) SPECTRAL TECHNIQUES Subject Code: MSPC-105

M.Code: MSPC-

Time: 3 Hrs. Max. Marks: 80

INSTRUCTIONS TO CANDIDATES: Attempt any FIVE questions out of EIGHT questions.

- 2. Each question carries SIXTEEN marks.
- 1. a) Describe the various types of electronic transitions observed in organic compounds when expose to UV& visible lights.
 - b) Explain various factor affecting wavelength maxima with example.
- 2. a) Explain the principle of I.R. Spectroscopy.
 - b) Explain dispersive spectrophotometer.
- 3. a) How can you distinguish between the following pairs of structures on the basis of PMR spectroscopy:
 - a. CH2Br-CH2Br & CH3-CHBr
 - b. CH3CH2OH & CH3-O-CH3
 - c. CH3-C6H5-CH3 & C6H5-CH2-CH3
 - d. CH3CH2CHO & CH3-CO-CH3
 - b) Explain Spin-Spin coupling and cause of splitting.
- 4. a) Explain Nitrogen Rule.
 - b) Explain different types of Mass Spectroscopy.
- 5. a) Explain Pharmaceutical application of mass spectroscopy.
 - b) Explain HOMCOR & HETCOR.

1 | M - 2 0 5 0 3 (S11) - 2 3 8

- 6. a) Explain the application of U.V. spectroscopy.
 - b) Explain effect of hydrogen bonding in vibrational frequency and vibrational coupling in I.R.
- 7. What is Chemical shift? Explain various factors affecting chemical shift.
- 8. a) Why does U.V. spectrum gives to absorption bands rather than sharp peaks?
 - b) Explain molecular vibrations in I.R. spectroscopy.

NOTE : Disclosure of Identity by writing Mobile No. or Making of passing request on any page of Answer Sheet will lead to UMC against the Student.

2 | M - 2 0 5 0 3 (S11) - 2 3 8