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Total No. of Pages : 02

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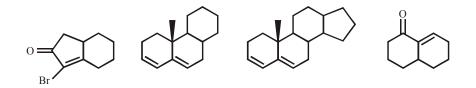
M.Sc. (Ph. Chem) (Sem.–1) SPECTRAL TECHNIQUES Subject Code : MSPC-105 M.Code : 20503 Date of Examination : 19-01-23

Time: 3 Hrs.

Max. marks : 80

INSTRUCTIONS TO CANDIDATES :

- 1. Attempt any FIVE questions out of EIGHT questions.
- 2. Each question carries SIXTEEN marks.
- 1. a) What do you mean by transition of electronic states? Write about different type of electronic transition of UV-visible spectroscopy. Why σ to σ^* transition is forbidden transitions. (8)
 - b) Describe the construction and working of photomultiplier tube. (8)
- 2. Define fundamental vibration and Fermi resonance. Describe in detail about the double beam IR spectrophotometer. (16)
- 3. a) Calculate the λ max of the following using wood-ward rules. (8)



b) Discus the major IR peaks to be observed in the spectrum of $CH_3(CH_2)_8CH_3$.

(8)

- 4. a) What are fragmentations in Mass spectrometry? Discuss the principle and instrumentation of mass spectrophotometer. Also write about basic fragmentation patter of MS. (12)
 - b) Write a note on chemical shift and coupling constant. (4)
- 5. a) Discuss the principle, instrumentation and working of continuous wave NMR instruments. (10)

| | b) Write a note on NOE. | (6) |
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| 6. | Define Hyphenation. Write the principle and instrumentation of GC-type of fragmentation techniques to use in GC-MS instruments. | -MS. Discuss various (16) |
| 7. | a) Elaborate, how solvent can affect the UV spectrum of an compou | unds? (4) |
| | b) Comment on various type of vibrations observed in IR spectrum. | (8) |
| | c) Write about the IR spectrum of anhydrides using suitable example | le. (4) |
| 8. | a) What is relaxation of nuclei. Elaborate, different relaxation provide NMR and their importance in ¹³ C NMR. | rocesses observed in (8) |
| | b) Describe various factors affecting chemical shift of a nuclei. | (8) |

NOTE : Disclosure of Identity by writing Mobile No. or Making of passing request on any page of Answer Sheet will lead to UMC against the Student.