Roll No. Total No. of Pages: 01

Total No. of Questions: 08

M.Sc. (Ph.Chem.) (Sem.-1)

ANALYTICAL TECHNIQUES - I

Subject Code: MSPC-103 M.Code: 20502

Date of Examination: 17-01-23

Time: 3 Hrs. Max. Marks: 80

INSTRUCTIONS TO CANDIDATES:

- 1. Attempt any FIVE questions out of EIGHT questions.
- 2. Each question carries SIXTEEN marks.
- 1. a) What is solubility product? Discuss the factors affecting the solubility of precipitates.
 - b) Write the calibration of IR and HPLC.
- 2. Write a note on:
 - a) Common ion effect
 - b) Method of expressing concentration
 - c) Primary and secondary standard
 - d) Standard deviation
 - e) Buffer and Buffer capacity.
- 3. Give the details of principle, methodology and application of Non aqueous titrations.
- 4. Give the details of principle, methodology and application of paper chromatography.
- 5. What are Redox Titrations? Write about various redox indicators. Elaborate on iodometric and iodimetric titrations.
- 6. Write the principle, instrumentation, and application of Gas chromatography with special reference to detection method. Give examples for derivatization GC method.
- 7. Give the details of principle, methodology and application of complexometric titrations.
- 8. What is Chromatography? Give its principle and classification. Write about various theories of chromatographic separations.

NOTE: Disclosure of Identity by writing Mobile No. or Making of passing request on any page of Answer Sheet will lead to UMC against the Student.

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