| Roll No. | | | | | | | |
|----------|--|--|--|--|--|--|--|
| | | | | | | | |

Total No. of Pages: 03

Total No. of Questions: 18

B.Tech. (2012 to 2017) (Sem.-1, 2) ELEMENTS OF MECHANICAL ENGINEERING

Subject Code: BTME-101 M.Code: 54101

Time: 3 Hrs. Max. Marks: 60

INSTRUCTIONS TO CANDIDATES:

- 1. SECTION-A is COMPULSORY consisting of TEN questions carrying TWO marks each.
- 2. SECTION B &C have FOUR questions each.
- 3. Attempt any FIVE questions from SECTION B& C carrying EIGHT marks each.
- 4. Select atleast TWO questions from SECTION B &C.

SECTION-A

Answer briefly / Fill in the blanks:

- 1) Define a Thermodynamics System.
- 2) Explain briefly zeroth law of thermodynamics.
- 3) What is Triple Point?
- 4) Define Internal Energy.
- 5) What do you mean by PMM of first kind?
- 6) Explain the concept of available and unavailable energy.
- 7) What is Air Standard Efficiency?
- 8) Define the term True Stress.
- 9) Under what conditions the center of mass and center of gravity coincide?
- 10) What is the difference between thermoplastic and thermosetting materials?

1 M-54101 (S1)-781

SECTION-B

- 11) To a close system 150kJ of work is supplied. If the initial volume is 0.6 m³ and the pressure of the system changes as p= 8-4V, where p is in bar and V is in m³, determine the final volume and pressure of the system.
- 12) 0.1 m³ of an ideal gas at 300K and 1 bar is compressed adiabatically to 8 bar. It is then cooled at constant volume and further expanded isothermally so as to reach the condition from where it started. Calculate:
 - a) Pressure at the end of constant volume cooling
 - b) Change in internal energy during constant volume process
 - c) Net work done and heat transferred during the cycle. Assume Cp = 4.3 kJ/kg K and Cv = 10.2 kJ/kg K.
- 13) Write down the general energy equations for steady flow system and simplify when applied to the following:
 - a) Centrifugal water pump
 - b) Steam nozzle
- 14) A closed system contains at a pressure 1 bar, temperature 300 K and volume 0.018 m³. This system undergoes a thermodynamics cycle consisting of the following three processes in series:
 - a) Constant volume heat addition till pressure becomes 5 bar,
 - b) Constant pressure cooling
 - c) Isothermal heating to initial state

Represent the cycle on T-S and p-V plots and evaluate the change in entropy for each process. Cp=0.718 kJ/kg K and R = 0.287 kJ/kg

2 | M-54101 (S1)-781

SECTION-C

- 15) An engine of 250 mm bore and 375 mm stroke works on otto cycle. The clearance volume is 0.00263 m³. The initial pressure and temperature are 1 bar and 50°C. If the maximum pressure is limited to 25 bar, find the following:
 - a) Air standard efficiency of the cycle
 - b) The mean effective pressure of the cycle, Assume the ideal conditions.
- 16) Discus briefly the method employed for improvement of thermal efficiency of open cycle gas turbines.
- 17) Discuss the following properties of the materials:
 - a) Ductility
 - b) Resilience
 - c) Weldability
 - d) Plasticity
- 18) Determine the moment of inertia of a semicircular arc about its diameter and hence locate its centroid.

NOTE: Disclosure of Identity by writing Mobile No. or Making of passing request on any page of Answer Sheet will lead to UMC against the Student.

3 | M-54101 (S1)-781