Roll No. Total No. of Pages: 02

Total No. of Questions: 09

B.Sc.(MLS) (Sem.-1)
BASIC OF BIOCHEMISTRY

Subject Code: BMLS-103-18

M.Code: 75260

Date of Examination: 14-01-2023

Time: 3 Hrs. Max. Marks: 60

INSTRUCTIONS TO CANDIDATES:

- SECTION-A is COMPULSORY consisting of TEN questions carrying TWO marks each
- 2. SECTION-B contains FIVE questions carrying FIVE marks each and students have to attempt any FOUR questions.
- 3. SECTION-C contains THREE questions carrying TEN marks each and students have to attempt any TWO questions.

SECTION-A

1. Answer briefly:

- a) Define S.I units.
- b) Give various types of solutions.
- c) Give importance of the pH.
- d) Define first aid.
- e) What is N/20 HC1 solution?
- f) Why should infected glass wares be sterilized before use?
- g) Define mole.
- h) What is the role of indicator in volumetric analysis?
- i) Define end point.
- j) For colorless solutions which meniscus should be observed?

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SECTION-B

- 2. Why the cleaning of glass-wares is very important?
- 3. Explain the working and principles of pH meter.
- 4. Explain the basic technique of storage of chemicals.
- 5. Write short note on care and maintenance of analytical balance.
- 6. What are S.I. and derived units and give S.I. units of the litre, Gram, mole and meter.

SECTION-C

- 7. Discuss about Henderson Hasselbalch equation and pH indicator.
- 8. Write about normal and molar solutions. How is standard solution prepared?
- 9. Explain the method of preparation of distilled water and its storage.

NOTE: Disclosure of Identity by writing Mobile No. or Making of passing request on any page of Answer Sheet will lead to UMC against the Student.

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